

Use of isotopes in metabolomics

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03-02-18

Synopsis

- **Natural abundance isotopes**
 - The value of the M+2 ion
- **Tracing a metabolic pathway**
 - ^{14}C -labeling a precursor for qualitative analysis
- **Isotope ratio outlier analysis**
 - Discerning real metabolites
 - Identifying metabolites
 - Use in quality control/quality assurance

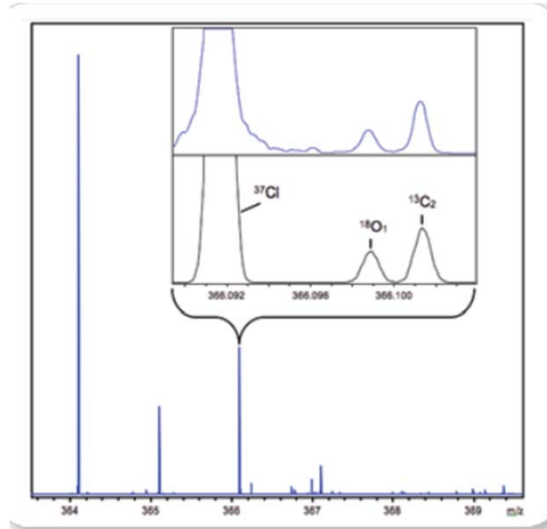
Value of natural isotopes

- The natural abundance of isotopes enables the investigator to determine the charge state of an ion
 - The principal contribution to $[M+H]^+$ or $[M-H]^-$ isotope ions comes from ^{13}C (~1.1% of all carbon atoms)
 - The intensity of the ^{13}C isotope ion increases relative to the number of carbon atoms
 - There is often an observable $^{13}\text{C}_2$ isotope peak

Value of the $[M+/-H+2]$ peak

- The mass difference due to a nominal increase in mass of 2 contains a lot of information
 - These are isotopic mass differences for each of the common elements
 - $^1\text{H}_2/{}^2\text{H}_2$ 2×1.006277 = 2.012554 (0.012%)
 - $^{12}\text{C}_2/{}^{13}\text{C}_2$ 2×1.003355 = 2.006710 (1.078%)
 - $^{14}\text{N}_2/{}^{15}\text{N}_2$ 2×0.997035 = 1.994079 (0.364%)
 - $^{16}\text{O}_2/{}^{17}\text{O}_2$ 2×1.004217 = 2.008434 (0.038%)
 - $^{16}\text{O}_2/{}^{18}\text{O}_1$ 1×2.004246 = 2.004246 (0.205%)
 - $^{32}\text{S}_2/{}^{33}\text{S}_2$ 2×0.999387 = 1.998774 (0.752%)
 - $^{32}\text{S}_2/{}^{34}\text{S}_1$ 1×1.995796 = 1.995796 (4.252%)

The importance of the M+2 ion



From Bruker

Using isotopes to trace a pathway

- Early studies (1930s) used ^2H , ^{13}C and ^{15}N labeling to map pathways
 - Limited to 1-200 m/z mass range
- 1950s/60s ^{14}C -radiotracers
 - 2D-Paper or thin layer chromatography
 - Radio gas chromatography
 - labeling of specific carbon atoms

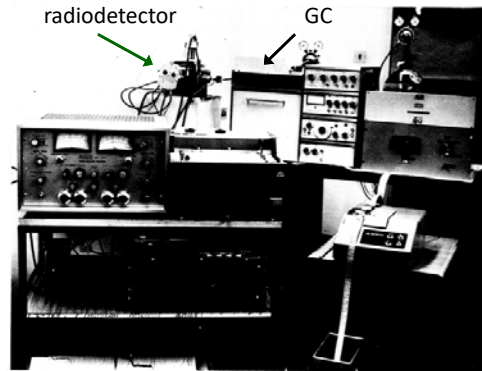
Origins of practical metabolomics Imperial College 1967-1970



Radio 2D-paper chromatography scanner
with digitization of collected data

The room had 20 of these scanners – data
analyzed by a central computer (in 1968)

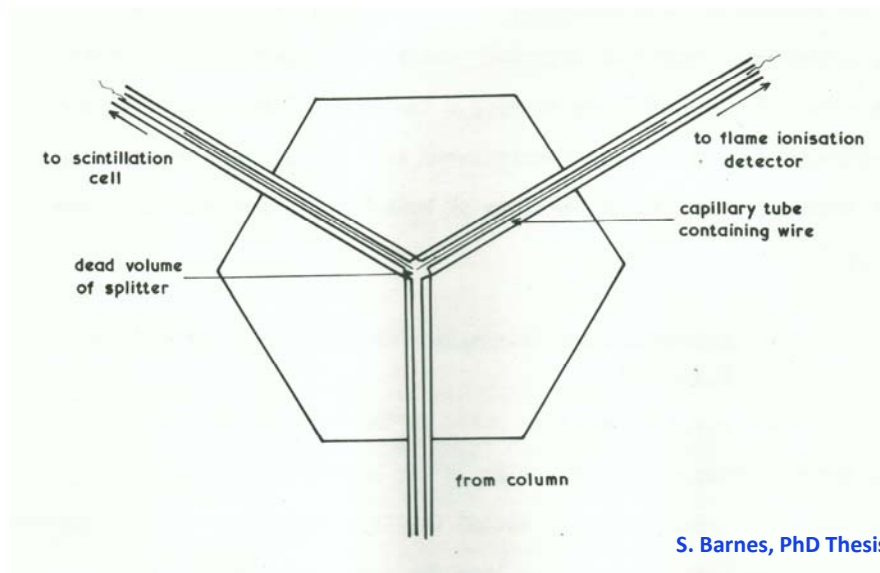
Courtesy of K.R. Mansford, PhD



Radio gas-liquid chromatography with
digitization of collected data

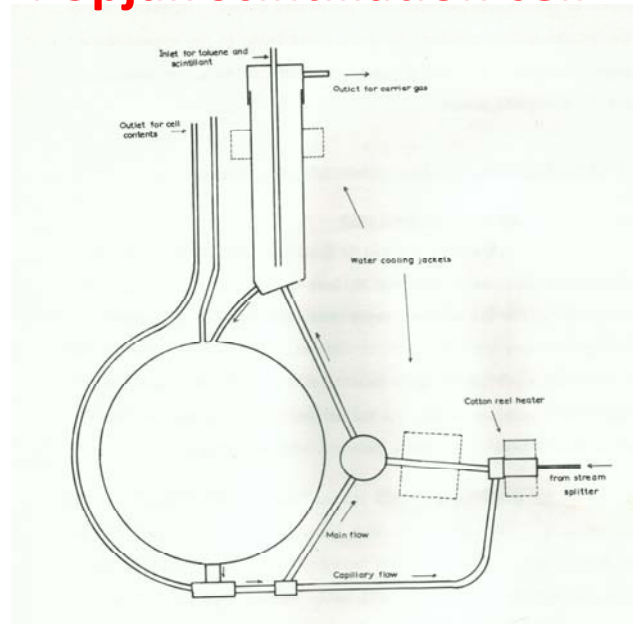
Developed this for my PhD work (1967-1970) to
study glucose metabolism in acellular slime moulds

Stream splitter for radio GC

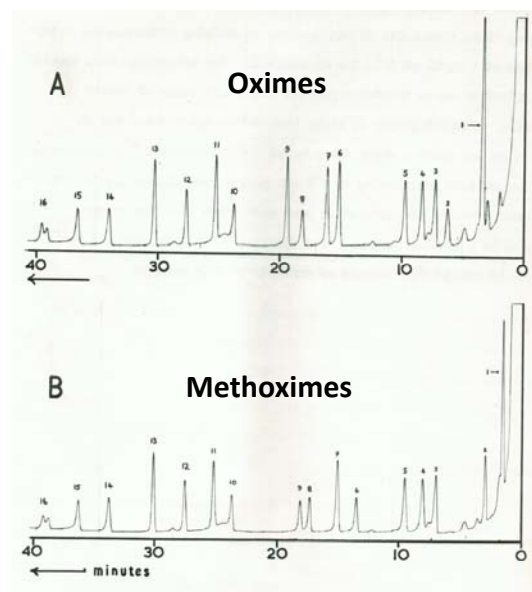


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Popjak scintillation cell



GC of glycolytic and Krebs cycle intermediates

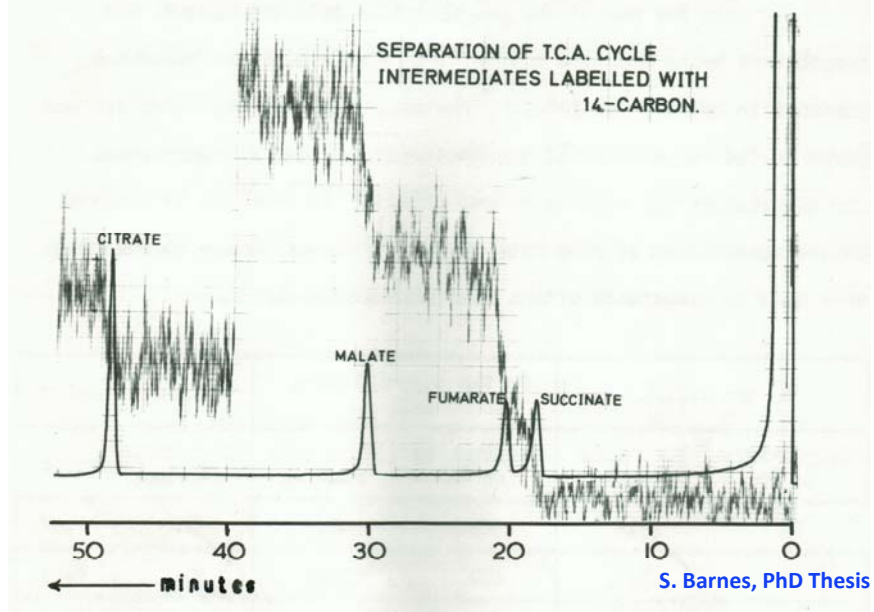


Temperature programming
of TMS ester/ethers on a 5' x
¼ inch packed column of
Chromosorb W coated with
OV-1 liquid phase

1=pyruvate, 2=?? ,
3=phosphate, 4=succinate,
5=fumarate, 6=oxaloacetate,
7=malate, 8=α-KG,
9=hexadecane, 10=α-GP,
11=citrate, 12=α-D-glucose,
13=β-D-glucose, 14=docosane,
15=F6P, 16=G6P

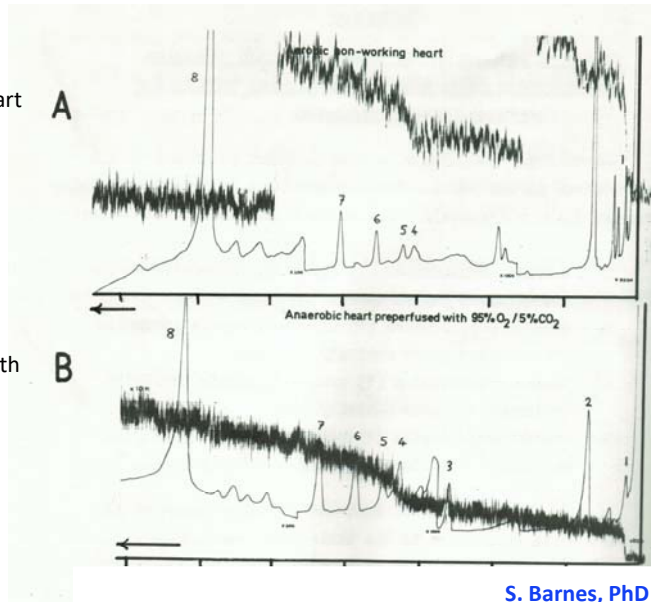
S. Barnes, PhD Thesis

Radio-GC of Krebs Cycle intermediates



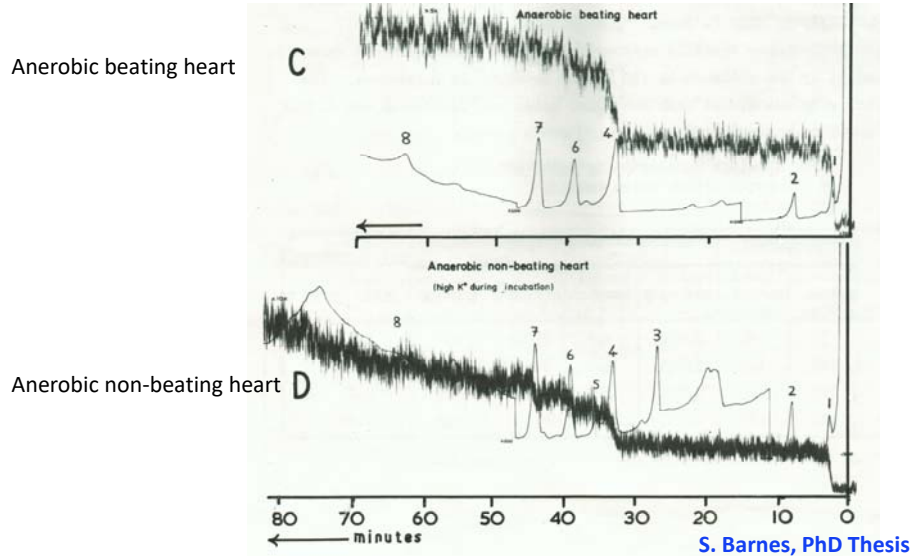
Radio GC analysis of beating heart

Aerobic beating heart



Anerobic beating
heart perfused with
95% O₂/5% CO₂

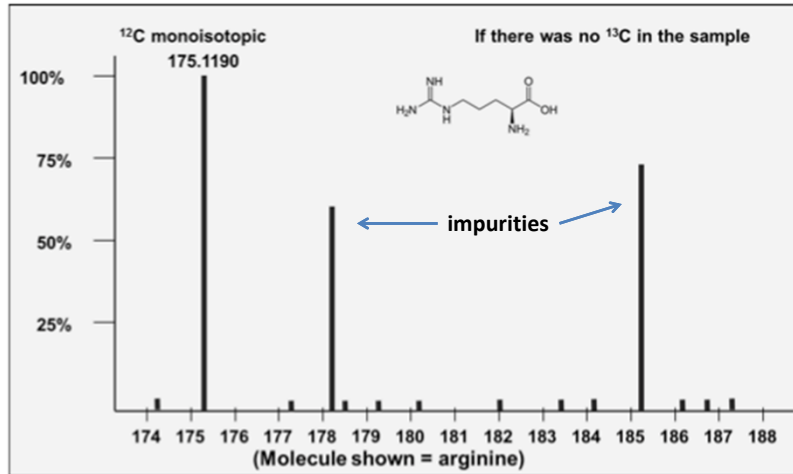
Radio GC analysis of anerobic heart



Tracking metabolites with IROA

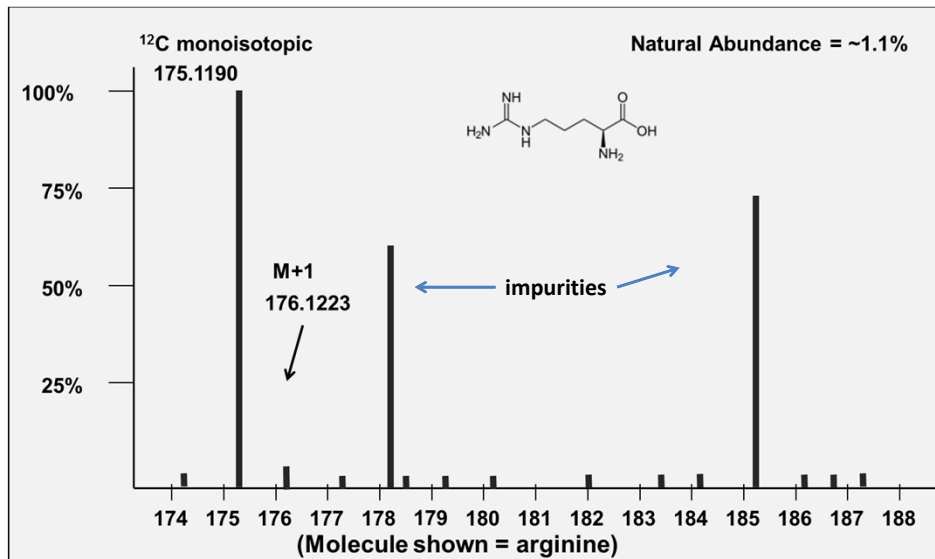
- Isotope ratio outlier analysis (IROA)
 - Not used for flux analysis, but rather to create a unique signal for metabolites
 - Used for LC-MS (and possibly GC-MS)
 - Designed to distinguish between metabolites of interest and background signals
 - Requires uniform labeling at the 95% and 5% ^{13}C -enrichment levels

All ^{12}C in arginine $[\text{M}+\text{H}]^+$



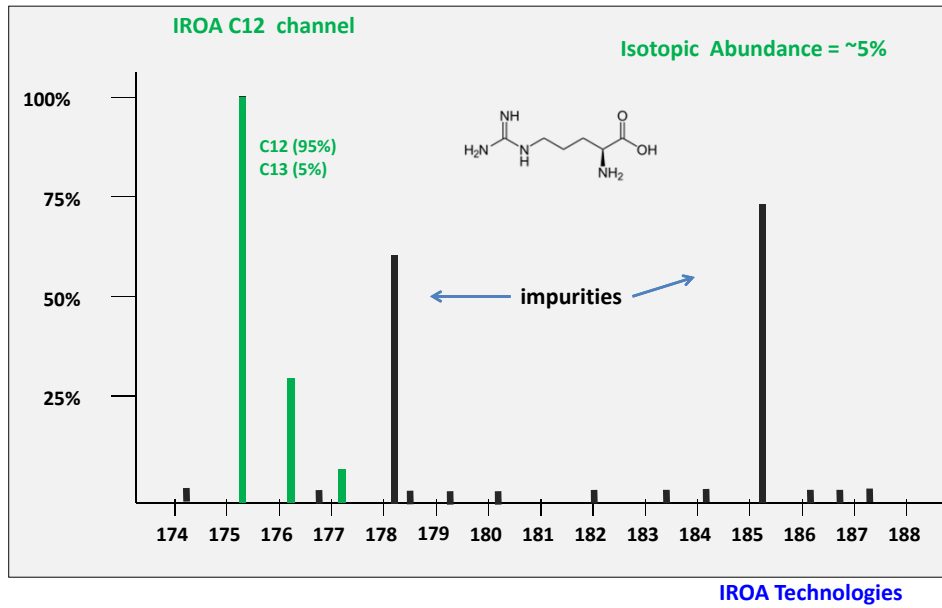
IROA Technologies

Natural abundance of ^{13}C in arginine

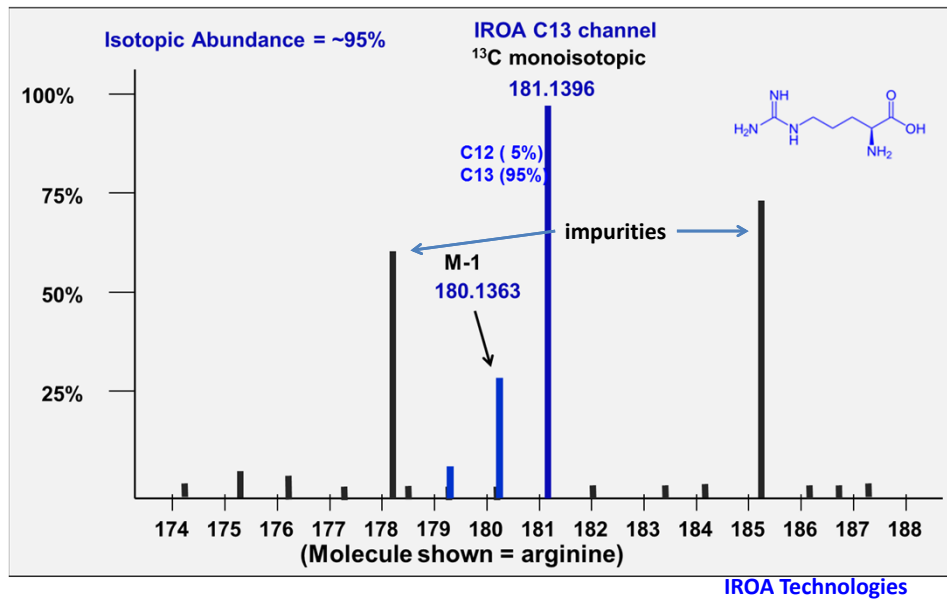


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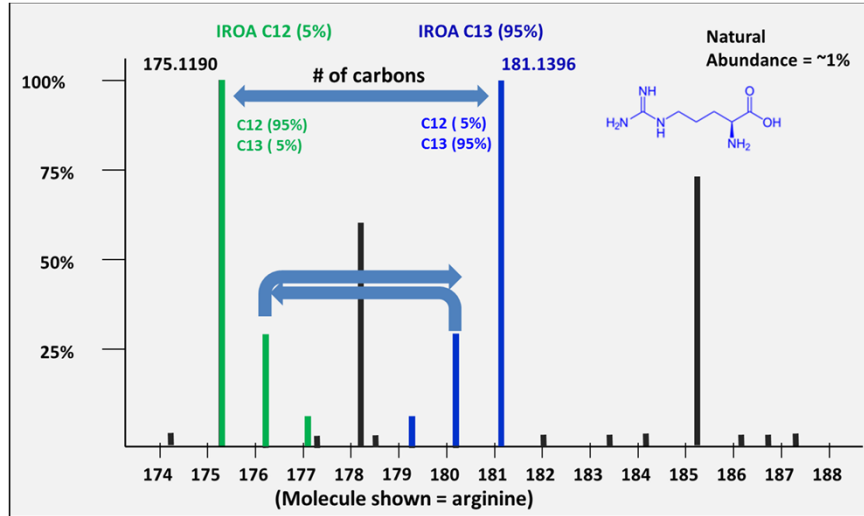
Making ^{13}C abundance = 5%



Making ^{13}C abundance = 95%

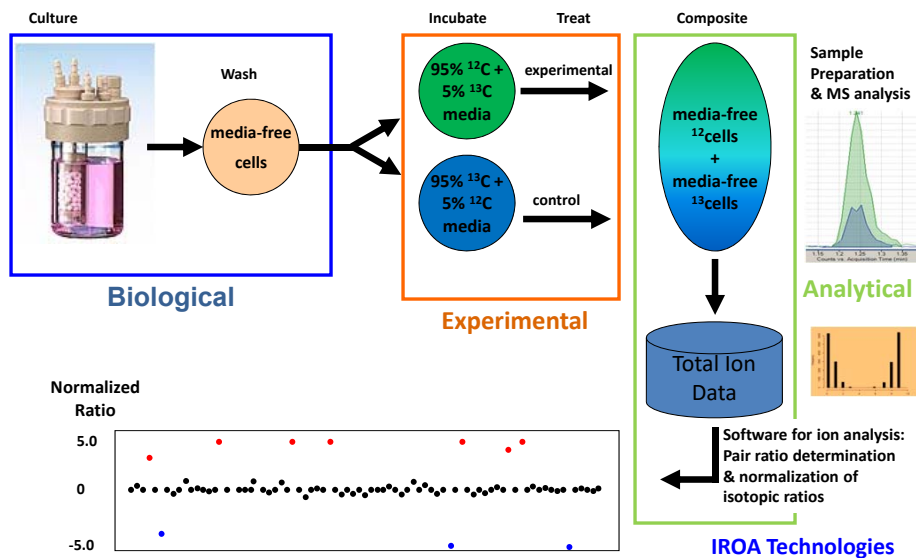


Pairing the 5% and 95% ¹³C-labeling



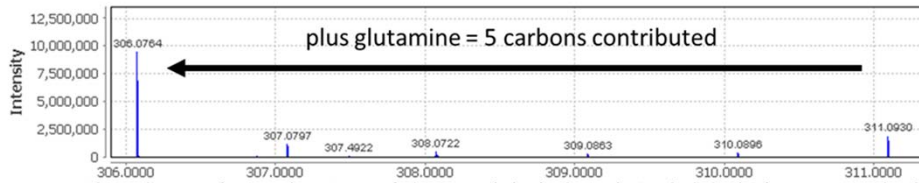
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The IROA approach

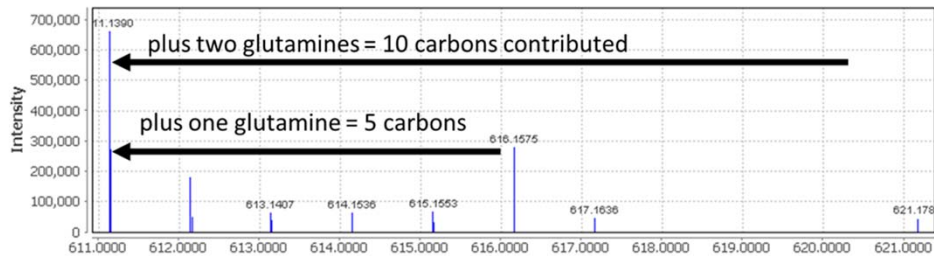


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The span of isotopes = # carbon atoms



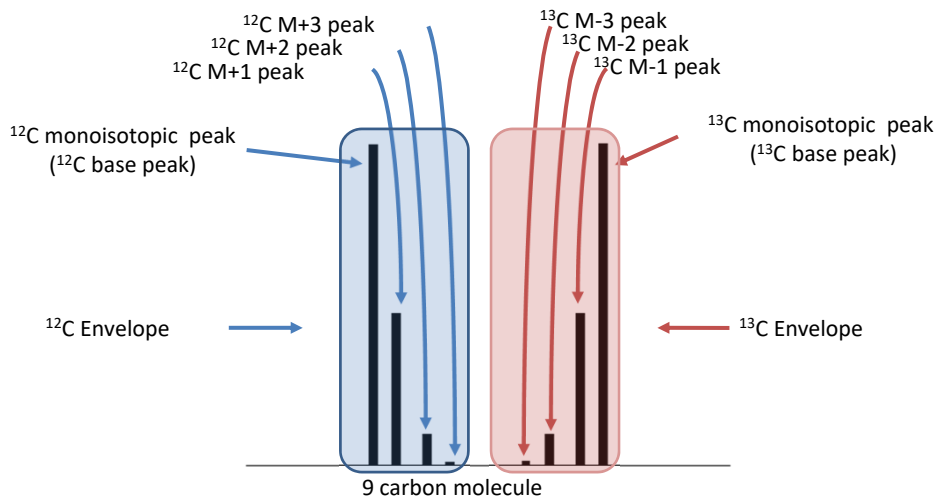
Glutathione shows the sizes of the non-labeled, and singly labeled species, which inherits a five carbon fragment from glutamine.



Oxidized glutathione shows the sizes of the non-labeled, singly labeled, and doubly labeled species. Each inheriting a five carbon fragment from glutamine.

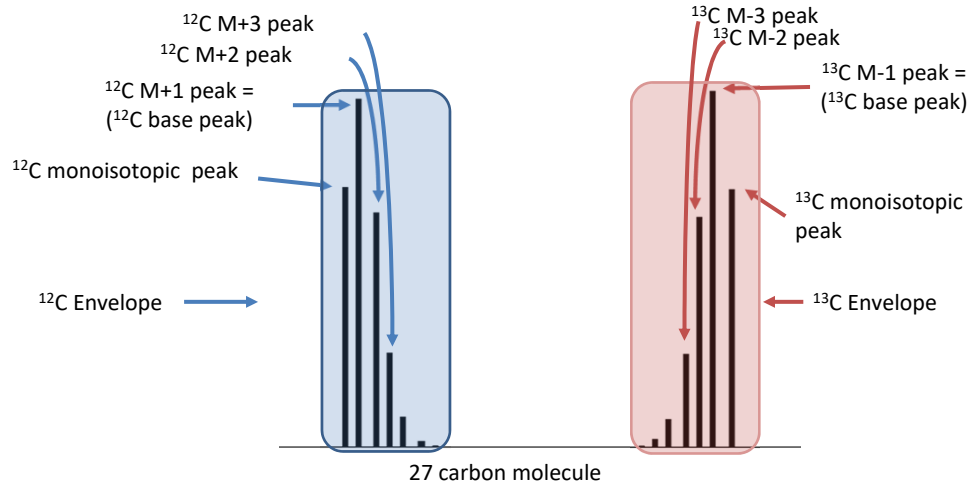
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The IROA peak pattern



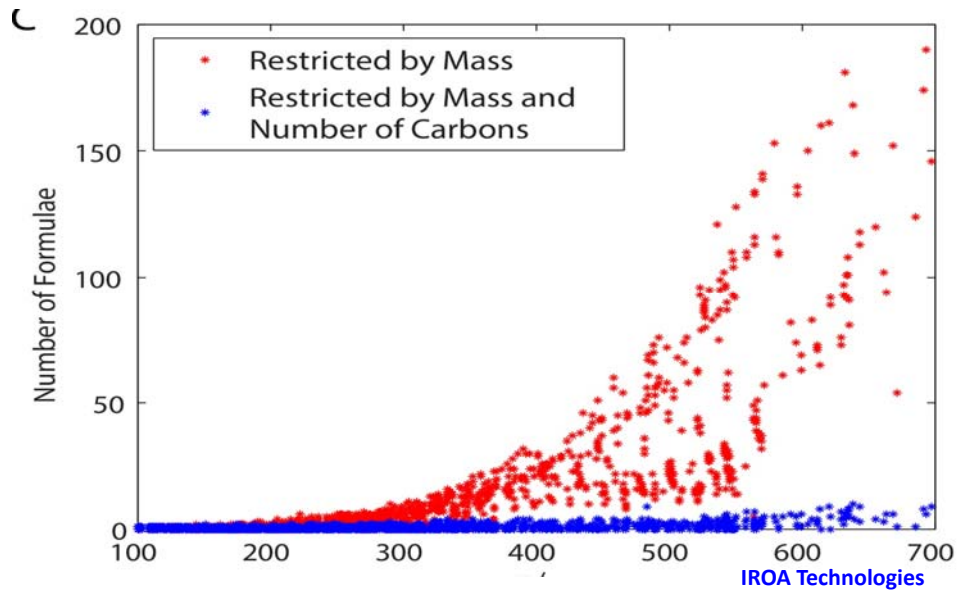
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IROA profile of a bigger metabolite

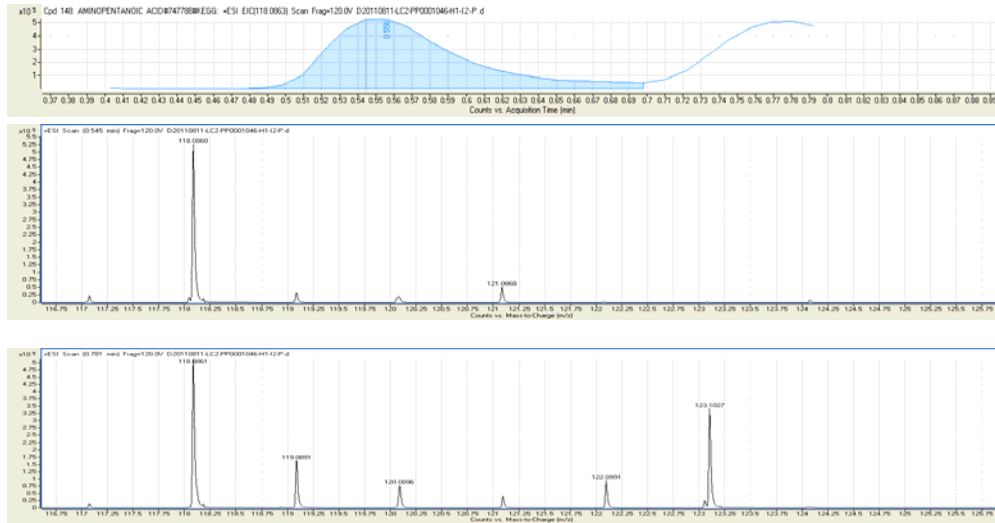


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Value of knowing the carbon

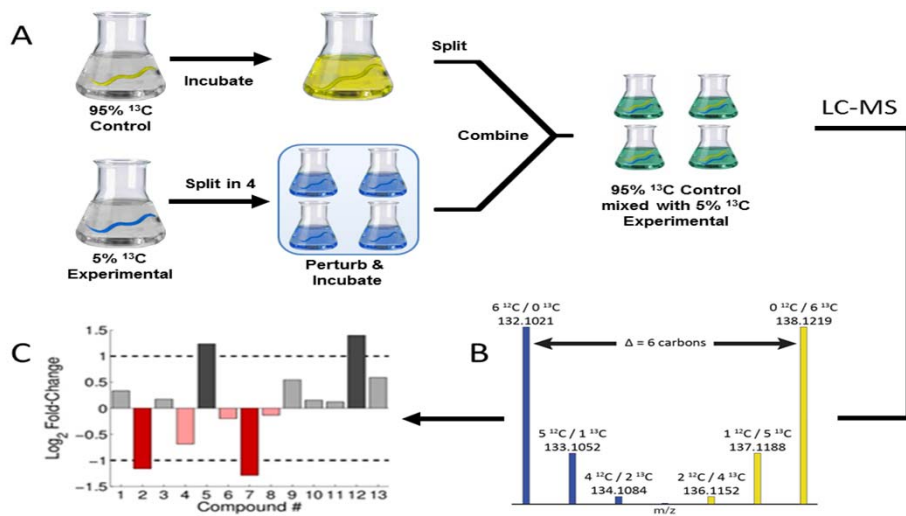


Avoiding metabolite artifacts with IROA



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IROA with *C. elegans*

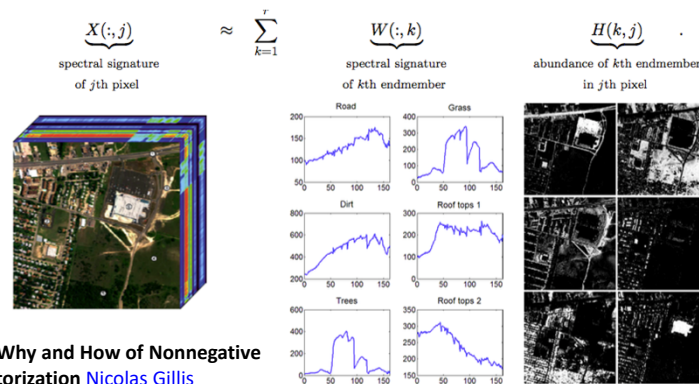
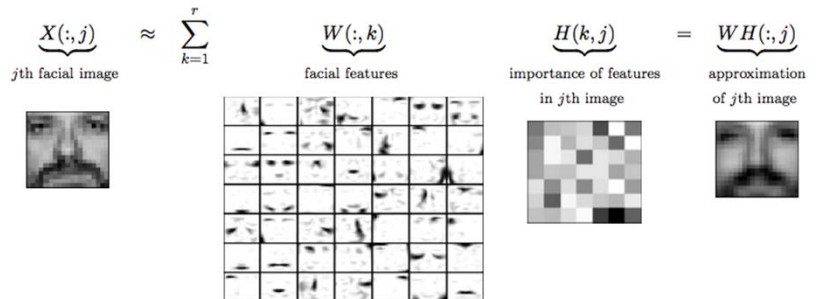


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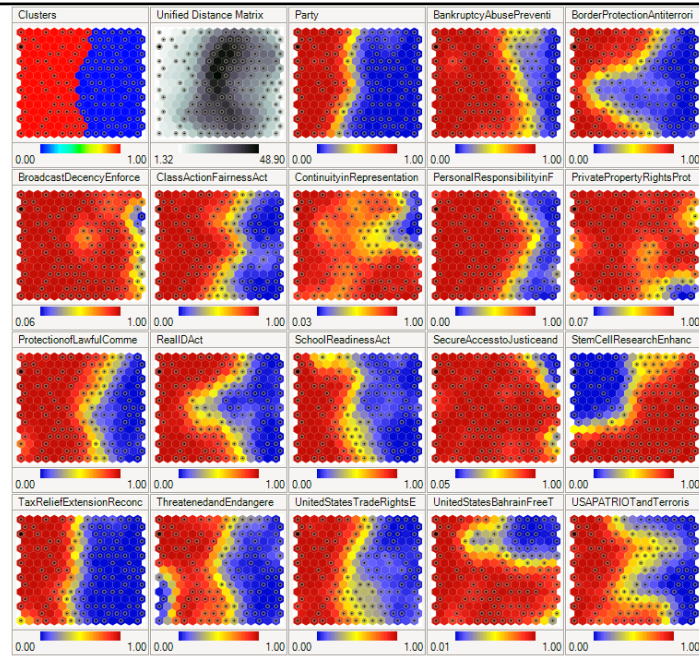
Effect of a toxin on *C. elegans*

- 742 strong IROA peak pairs were found
 - 314 named / 428 formula determined
 - Positive and negative mode LC
 - Thermo Orbi-trap @ 70K resolution
- Strong response signature determined
 - Basic statistics, PCA, Random Forest, non-negative matrix factorization (NMF), self-organizing map (SOM)
 - 74 compounds were considered significant by at least 3 of these methods.

IROA Technologies



From: [The Why and How of Nonnegative Matrix Factorization](#) [Nicolas Gillis](#)



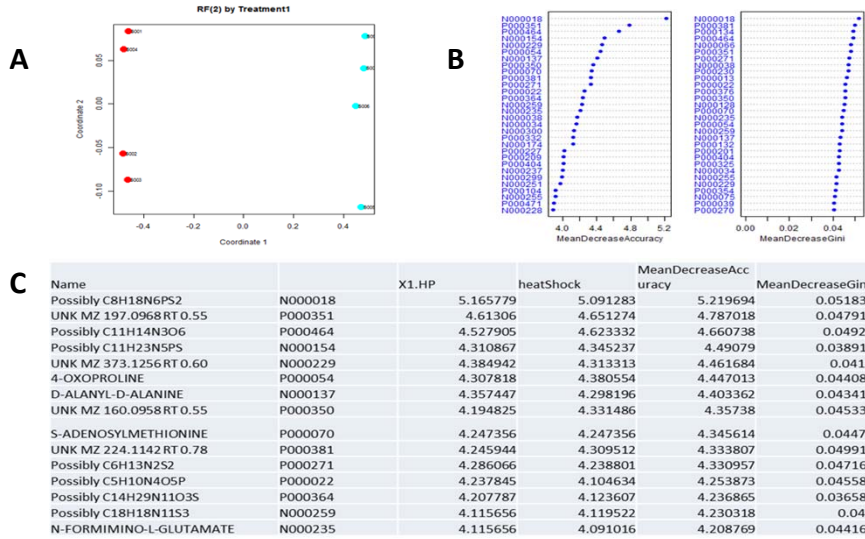
https://en.wikipedia.org/wiki/Self-organizing_map#/media/File:Synapse_Self-Organizing_Map.png

Ions significantly affected by the toxin

Name		p.value	F-Value treatment ²
L-LYSINE	P000009	7.89E-05	89.71
Possibly C ₇ H ₇ N ₃ O ₂ S	P000018	3.06E-05	124.99
L-ARGININE	P000019	0.000131	74.84
Possibly C ₆ H ₉ NO ₁₁	P000025	0.000182	66.63
UNK <i>m/z</i> 369.2215 RT 0.58	P000040	2.19E-05	140.24
SACCHAROPINE	P000046	7.23E-05	92.51
L-THREONINE	P000051	2.64E-05	131.52
L-GLUTAMIC ACID	P000053	1.09E-06	389.79
4-OXOPROLINE	P000054	1.74E-05	151.81
Possibly C ₄ H ₇ NO	P000058	1.8E-05	150.26
L-VALINE	P000060	0.000262	58.37
CITRULLINE	P000061	3.15E-05	123.67
4-METHYLENE-L-GLUTAMINE	P000062	0.000169	68.40
L-METHIONINE S-OXIDE	P000065	7.55E-06	202.32
L-PROLINAMIDE	P000085	0.000227	61.56
STACHYDRINE	P000102	4.75E-05	107.19
UNK <i>m/z</i> 206.0368 RT 0.71	P000114	0.000251	59.35
N-ACETYLPUTRESCINE	P000122	8.96E-07	417.06
EPSILON-CAPROLACTAM	P000123	1.29E-08	1731.72
2-AMINO-OCTANOIC ACID	P000131	0.000213	62.99
UNK <i>m/z</i> 345.1258 RT 0.97	P000141	0.000111	79.36
Possibly C ₁₀ H ₁₉ N ₂ O ₅ P ₂	P000151	0.000154	70.78
CYS-GLY	P000152	0.000116	78.29
URATE	P000156	0.000222	62.02
Possibly C ₁₃ H ₁₆ N ₆ OPS	P000218	1.1E-05	177.82

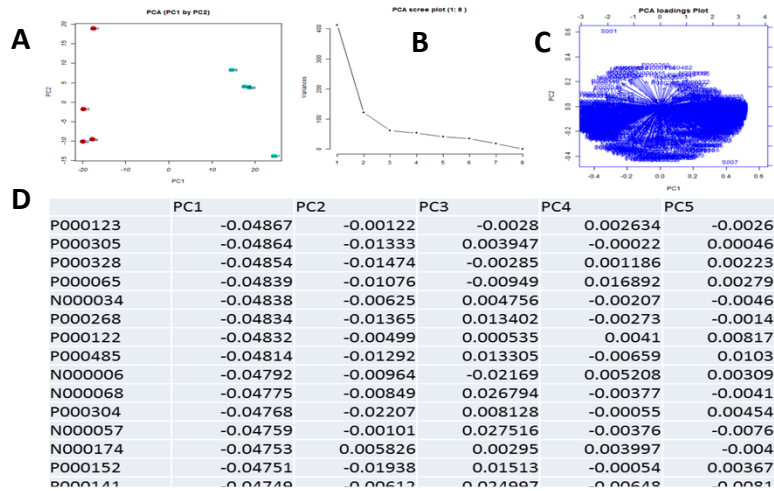
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Multivariate statistics



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PCA analysis of toxin's effect



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Summary of most likely metabolites

Name		Stats1	RFTop1	RFTop2	NMF3	NMF4	NMF5	NMF6	Count
UNK <i>m/z</i> 160.0958 RT 0.55	P000350	1	1	1	1	1	1	1	7
UNK <i>m/z</i> 197.0968 RT 0.55	P000351	1	1	1	1	1	1	1	7
UNK <i>m/z</i> 216.0852 RT 0.61	N000034	1	1	1	1	1	1	1	7
D-ALANYL-D-ALANINE	N000137	1	1	1	1	1	1	1	7
Possibly C ₂₃ H ₃₄ N ₂ O ₅	N000174	1	1	1	1	1	1	1	7
UNK <i>m/z</i> 373.1256 RT 0.60	N000229	1	1	1	1	1	1	1	7
2-AMINO-OCTANOIC ACID	P000131	1	1	0	1	1	1	1	6
Possibly C ₆ H ₈ N ₄ O ₃	P000354	1	1	0	1	1	1	1	6
UNK <i>m/z</i> 510.2122 RT 0.68	P000373	1	1	0	1	1	1	1	6
UNK <i>m/z</i> 224.1142 RT 0.78	P000381	0	1	1	1	1	1	1	6
Possibly C ₈ H ₉ NO ₄ P	P000410	1	1	0	1	1	1	1	6
Possibly C ₁₁ H ₁₄ N ₃ O ₆	P000464	0	1	1	1	1	1	1	6
Possibly C ₆ H ₈ N ₄ O ₃ P	P000471	1	1	0	1	1	1	1	6
Possibly C ₆ H ₁₂ N ₂ O ₅ PS	N000006	1	1	0	1	1	1	1	6
Possibly C ₁₁ H ₂₃ N ₃ PS	N000154	1	1	1	0	1	1	1	6
D-GLUCOSE	N000228	1	1	0	1	1	1	1	6
UNK <i>m/z</i> 548.2037 RT 0.63	N000232	1	1	0	1	1	1	1	6
GLYCERATE	N000237	1	1	0	1	1	1	1	6

IROA Technologies

From Chris Beecher, IROA Technologies

https://www.uab.edu/proteomics/metabolomics/workshop/2017/videos/beecher_day3.html

The IROA Workflow

- The IROA Workflow is an extension of the Phenotypic Protocol in which:
 - a defined IROA-based Internal Standard (IS) is used in any type of experimental or clinical sample, and
 - an equally defined QA/QC sample (Matrix) is analyzed daily.
- Together they make a systematic measurement system that is completely reproducible across sample types, instrument types, and overcomes time-induced variance.

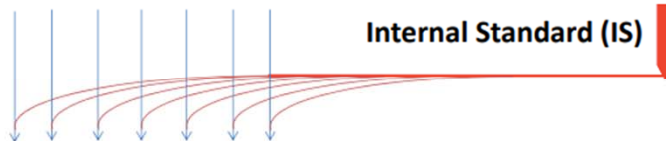
The IROA Workflow is the basis of “Clinical Metabolomics”.

IROA-based Workflow

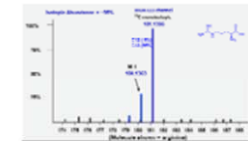
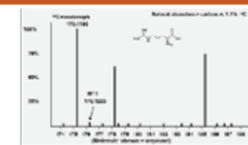
Experimental Samples



Internal Standard (IS)



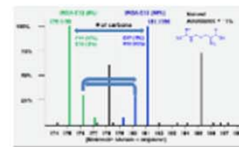
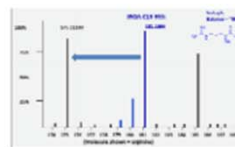
Experimental Samples with IS



+



Matrix

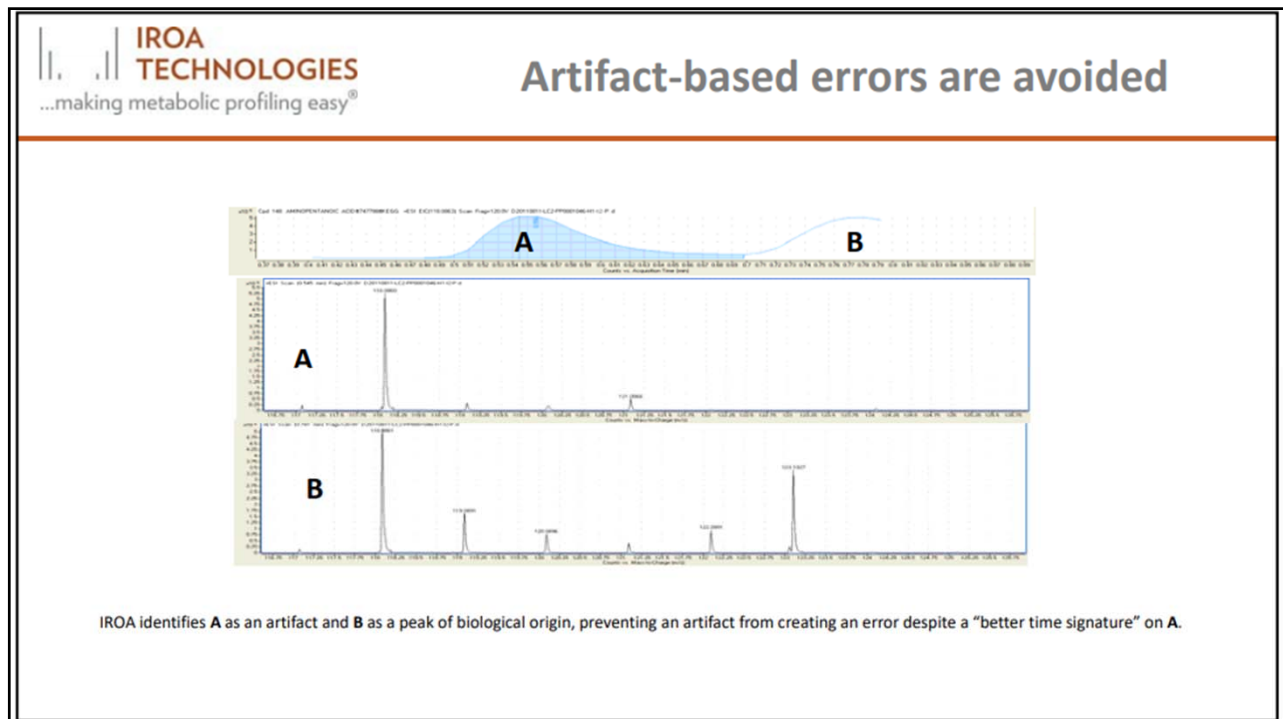
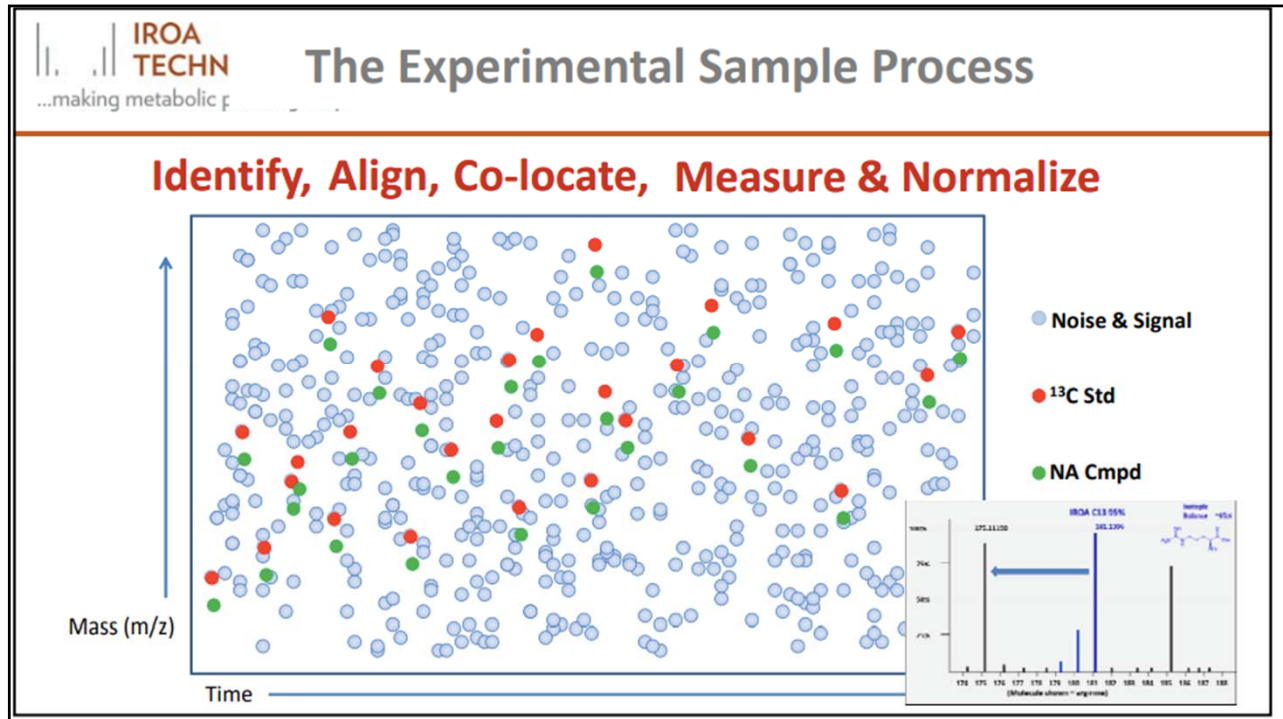


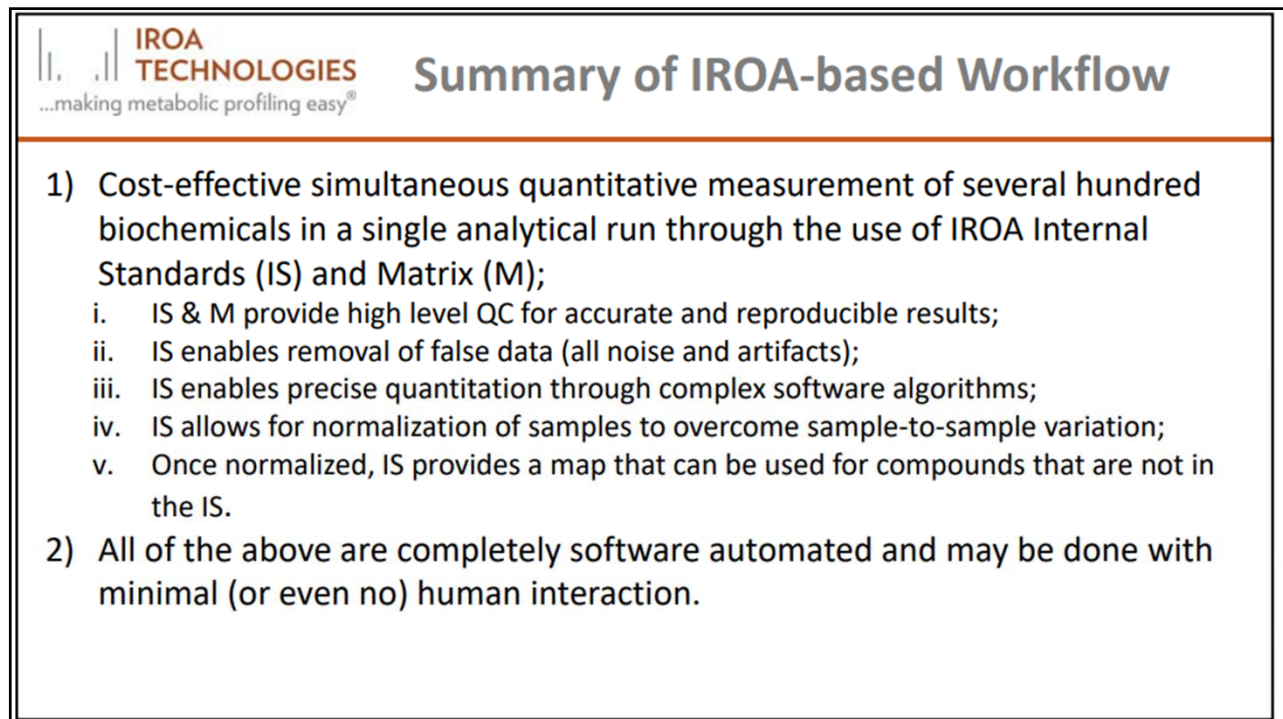
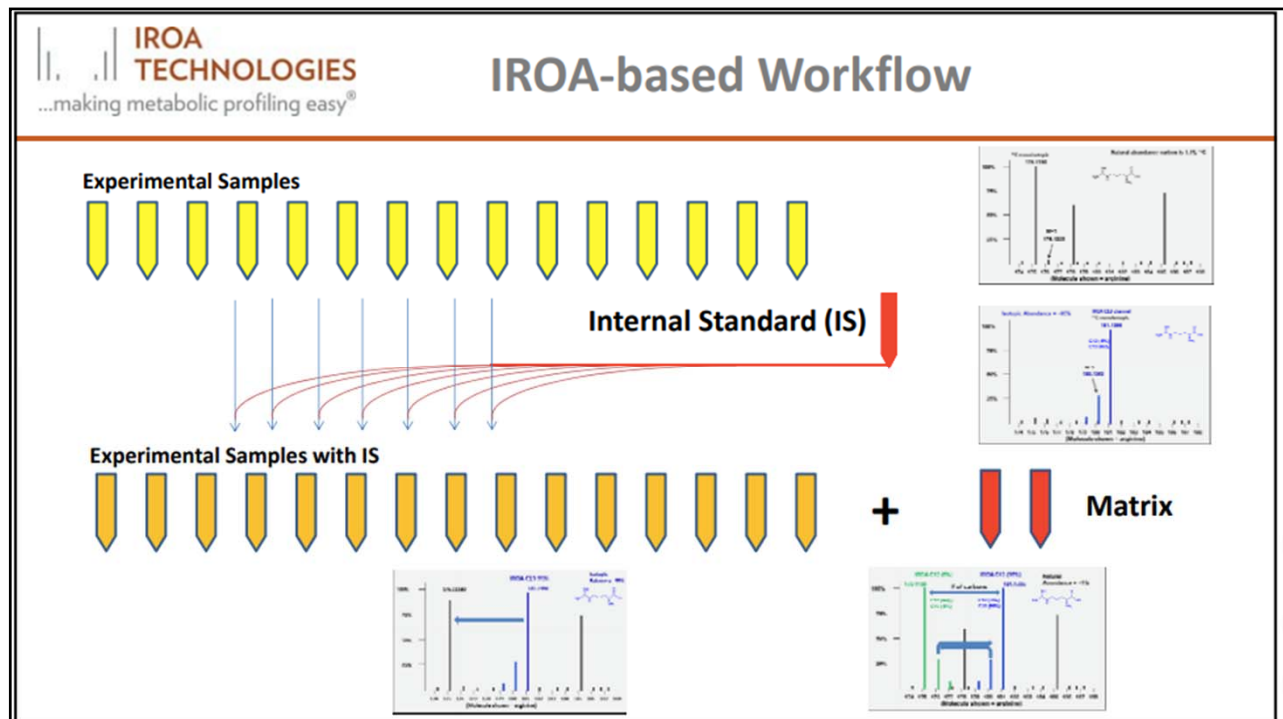
What is the IROA-IS

- The IS is a 95% U-¹³C-labeled complex Internal Standard
- The IS has a standard concentration of 1000+ **identified** and curated compounds for **co-location in an experimental sample**.
- The IS has enough compounds to provide for a Retention Time (RT) ladder that allows **alignment** of all peaks in the chromatogram.
- The IS may be used to **normalize** the samples against one another.
- The IS allows **day-to-day, or even instrument-to-instrument variances to be eliminated**.

What is Matrix?

- The Matrix sample is a made from the same material as the IS but rather than having a natural abundance partner it is paired with a perfectly matched IROA 5% U-¹³C sample.
 - The almost perfect balance of the 5% and 95% chemical composition,
 - The completely defined nature of the Matrix sample, and
 - The absolute reproducibility of the Matrix sample.
- Provide a way to compare day-to-day analytical performance on all parts of the analytical process, and
- Provide a daily mapping of all compounds found in the IS so that their complete identification is always assured.





The 1500 peaks in Matrix (pos mode)

While finding them is completely automated, we are currently examining each one, and annotating it

- We have built a database to collect all of this information
- We are using this to directly tackle the problem of the percent of peaks that are “knowns”, and what percent are “unknowns”, i.e. not just fragments, adducts, etc. of “knowns”

*We will be successful because the Matrix is a pure IROA mixture, i.e. we can discriminate between real compounds and artifacts

